



INSTITUTO DE
TECNOLOGÍA
QUÍMICA

Conversion and Storage of Energy

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EXCELENCIA
SEVERO
OCHOA



CSIC
CONSEJO SUPERIOR DE INVESTIGACIONES CIENTÍFICAS



UNIVERSITAT
POLITÈCNICA
DE VALÈNCIA

Enabling components and technologies for decarbonized industry

Energy Efficiency

CO₂ capture

Circularity

Electrification

H₂ production

Low C-footprint fuels



**Smart
Catalysts**

Electrochemical Cells

**Smart Electric
Heating**

**Smart
Separations**

**Gasification &
Pyrolysis**

**Heat
Exchangers**

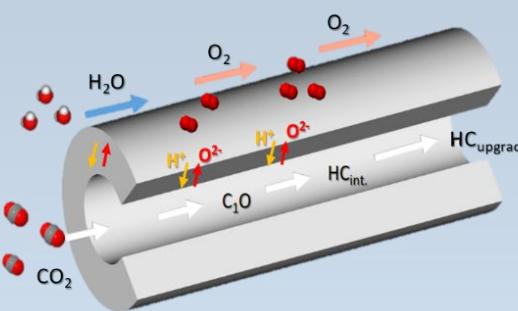
**Solar Heating
& Cycles**

Conversion and Storage of Energy

<https://itqmembranes.itq.webs.upv.es/>

Catalytic membrane reactors

Catalytic membrane reactors couple chemical reactions with membrane separation technology providing compact systems with improved performance (selectivities and/or yields). We pay special attention to the design of cutting-edge membrane reactors but also to test the reactors performance and long-term stability for the next generation of renewable fuel and chemicals production, and CO₂ valorization systems.

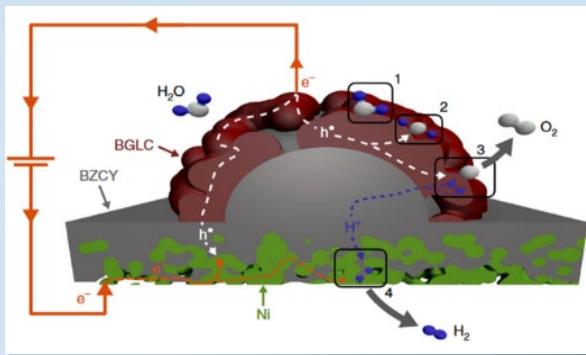


Membrane technology

We have wide expertise on the production and characterization of mixed ionic electronic conducting membranes at bench and prototype scales. These high temperature membranes would find application where high purity oxygen or hydrogen is required.

Electrochemical conversion & storage devices

Fuel cells and electrolyzers are electrochemical devices that directly convert/store the chemical energy into/from electrical energy. In particular, we focus on the fabrication and characterization of Solid oxide fuel/electrolyser cells (SOFC/SOECs) and Protonic ceramic fuel/electrolyser cells (PCFC/PCECs)



nature materials

ARTICLES
<https://doi.org/10.1038/s41563-019-0388-2>

Mixed proton and electron conducting double perovskite anodes for stable and efficient tubular proton ceramic electrolyzers

Einar Vollestad^{1,2}, Ragnar Strandbakke¹, Mateusz Tarach¹, David Catalán-Martínez²,

Marie-Laure Fontaine¹, Dustin Beaufort¹, Daniel R. Clark^{1,4}, Jose M. Serra³ and Truls Norby^{1*}

Hydrogen production from water electrolysis is a key enabling energy storage technology for the large-scale deployment of intermittent renewable energy sources. Proton ceramic electrolyzers (PCEs) can produce dry pressurized hydrogen directly from steam, avoiding major parts of cost-driving downstream separation and compression. However, the development of PCEs has been limited by the lack of a suitable anode material. Here we report a mixed protonic-electronic conductor (BZCY) that is fully operational Ba_{0.5}Zr_{0.1}O₃-based tubular PCE, with 10 cm² active area and a hydrogen production rate above 15 Nm³ min⁻¹. The total steam-to-hydrogen partial polarization resistances below 12 cm at 600 °C and Faradaic efficiencies close to 100% at high steam pressures. These tubular PCEs are mechanically robust, tolerate high pressures, allow improved process integration and offer scale-up modularity.

nature energy

ARTICLES
<https://doi.org/10.1038/s41560-020-00720-4>

Hydrogen production via microwave-induced water splitting at low temperature

J. M. Serra^{1,2*}, J. F. T. Blasco^{1,2}, B. García-Báños³, M. Balaguer¹, P. Plaza-González^{2,3},

J. Santos-Blasco^{2,3}, D. Catalán-Martínez^{2,3}, L. Navarrete^{2,3} and J. M. Catalá-Civera^{1,2,3}

Supplying global energy demand with CO₂-free technologies is becoming feasible thanks to the rising affordability of renewable resources. Hydrogen is a promising vector in the decarbonization of energy systems, but more efficient and scalable synthesis is required to enable its widespread deployment. Here we report contactless H₂ production via water electrolysis mediated by the microwave-triggered redox activation of solid-state ion-conducting CeO₂ that was previously electrochemically deoxydized by the sole reaction with the non-equilibrium gallium-doped CeO₂ that was previously electrochemically deoxydized by the sole reaction with the non-equilibrium gallium-doped CeO₂. This method identifies a new mechanism for water splitting catalysis and O₂ release. This process was cyclical, whereas H₂ yield and energy efficiency were material- and power-dependent. Deoxygenation of low-energy molecules (H₂O or CO₂) led to the formation of energy carriers and enabled CH₄ production when integrated with a Sabatier reactor. This method could be extended to other reactions such as intensified hydrocarbons synthesis or oxidation.

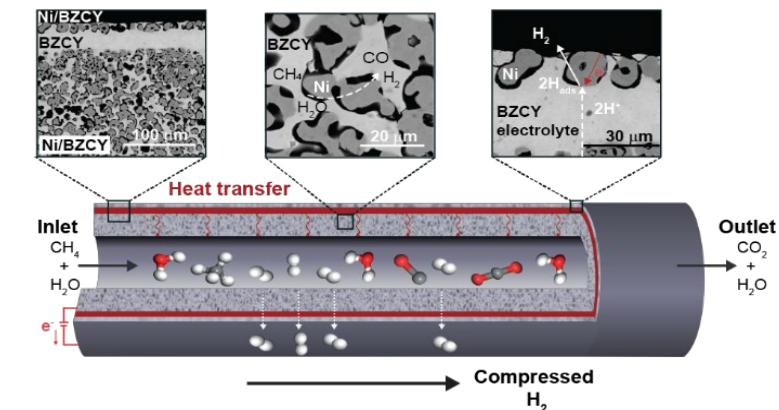
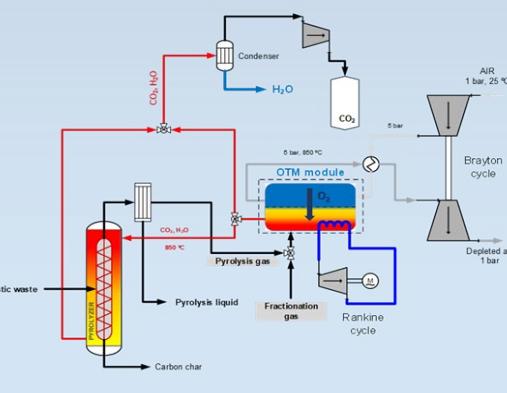
KERIONICS

RESEARCH
Clark et al., Science 376, 390–393 (2022) 22 April 2022

CATeGORY
Single-step hydrogen production from NH₃, CH₄, and biogas in stacked proton ceramic reactors

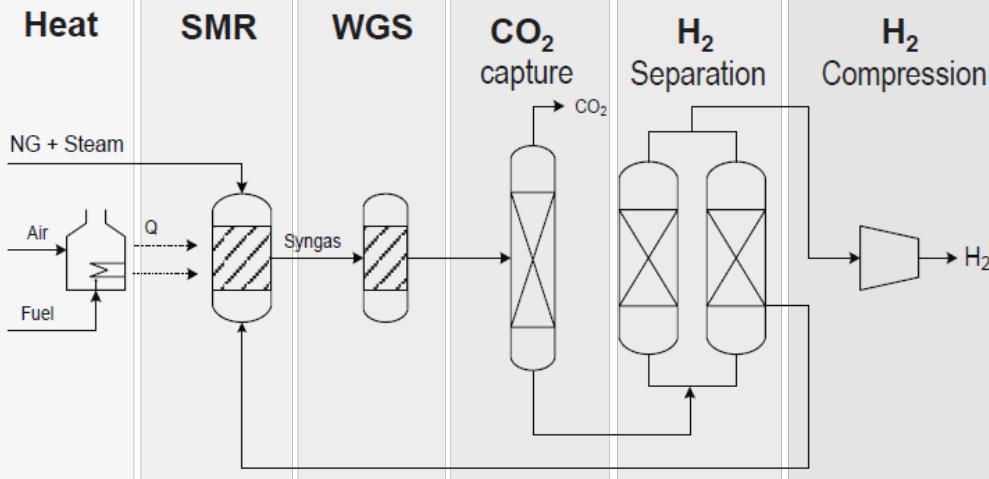
Daniel Clark¹, Harold Maledict-Fjeld², Michael Budd¹, Irene Yuste-Tirado^{1,2}, Dustin Beaufort¹, Simon Ameduri¹, Kevin Nguyen¹, Luca Amalio¹, Thijis Peters³, Per K. Vestre¹, Dimitrios K. Pappas¹, Jose M. Serra^{4*}, Sonja Remiro-Buenamahana⁵, Truls Norby¹, Tor S. Bjørheim¹

Proton ceramic reactors demand high temperatures to enable the conversion of ammonia, methane, and biogas to hydrogen. Hydrogen is a promising vector in the decarbonization of energy systems, but more efficient and scalable synthesis is required to enable its widespread deployment. Here we report contactless H₂ production via water electrolysis mediated by the microwave-triggered redox activation of solid-state ion-conducting CeO₂ that was previously electrochemically deoxydized by the sole reaction with the non-equilibrium gallium-doped CeO₂ that was previously electrochemically deoxydized by the sole reaction with the non-equilibrium gallium-doped CeO₂. This method identifies a new mechanism for water splitting catalysis and O₂ release. This process was cyclical, whereas H₂ yield and energy efficiency were material- and power-dependent. Deoxygenation of low-energy molecules (H₂O or CO₂) led to the formation of energy carriers and enabled CH₄ production when integrated with a Sabatier reactor. This method could be extended to other reactions such as intensified hydrocarbons synthesis or oxidation.

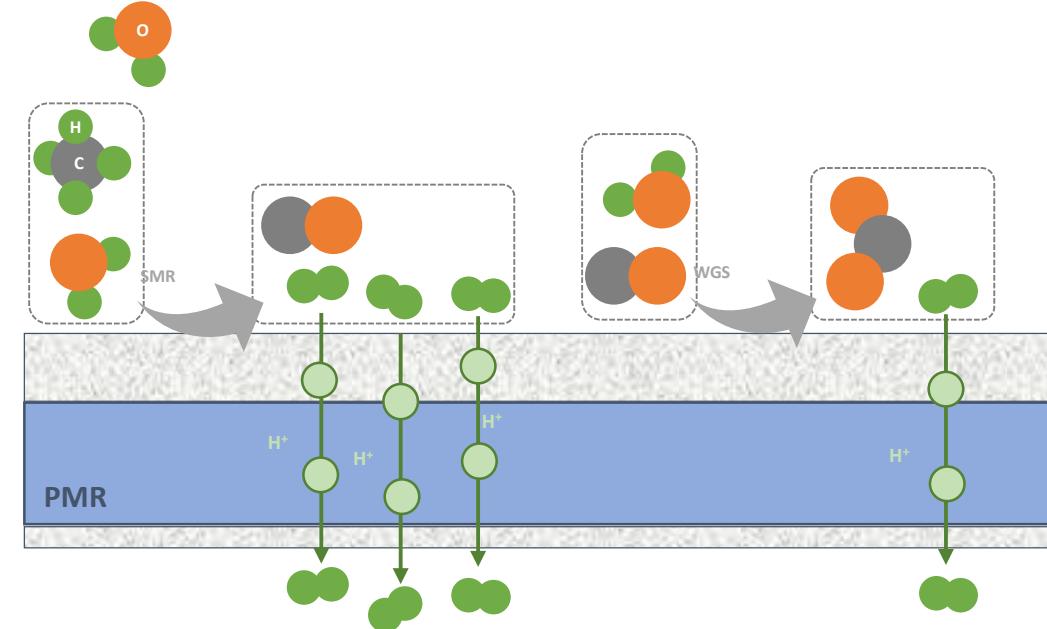


Electrified Membrane Reformer (eMR)

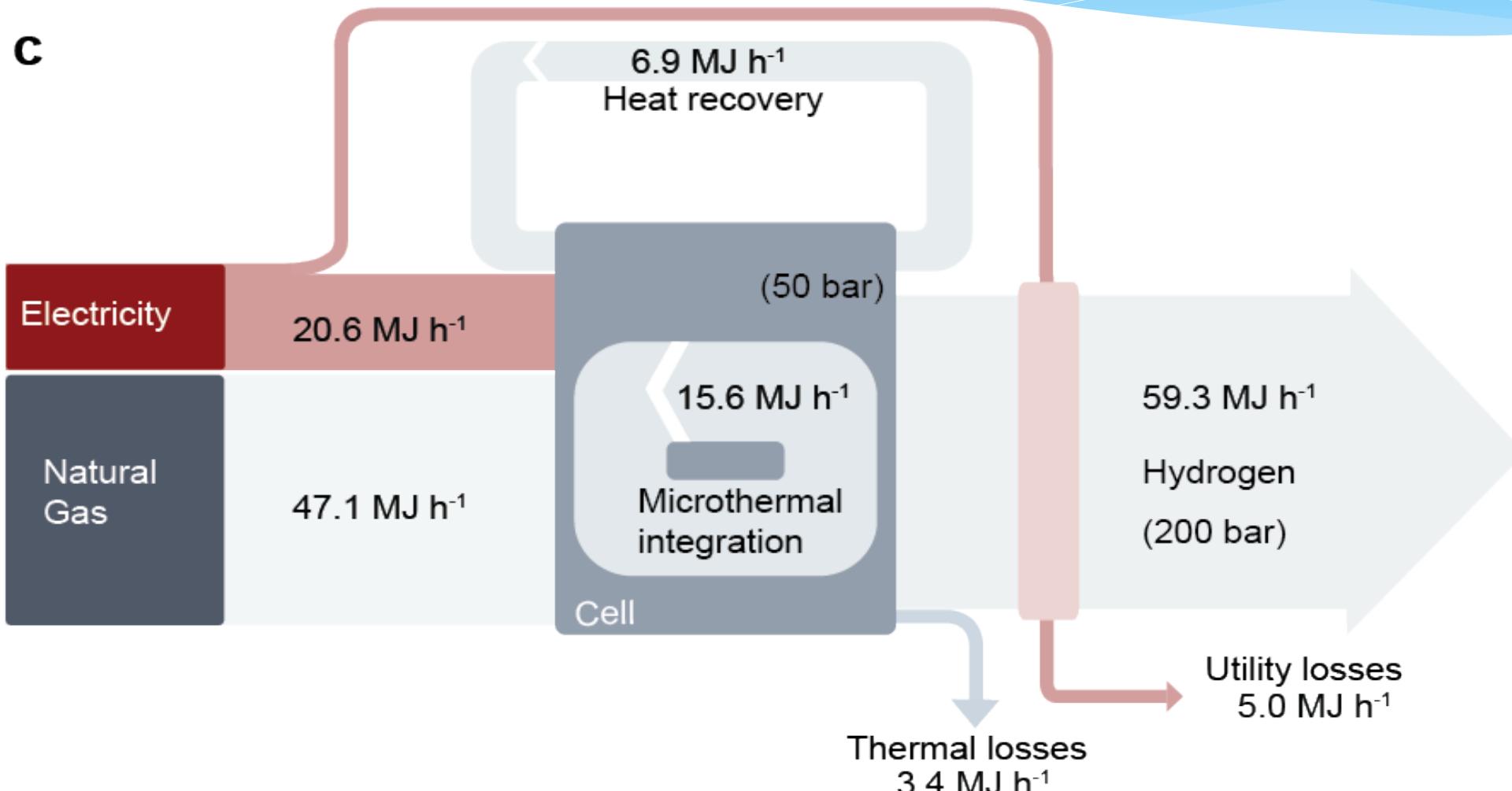
Steam Methane Reforming w. CCS (6 steps)



Process step reduction



Proton Membrane Reformer (PMR) – Energy Balance



H. Malerød-Fjeld et al., *Nature Energy*, 2, p. 923–931 (2017)



Proton Membrane Reformer (PMR) – Efficiency



**Hydrogen from natural
gas with carbon capture**

>90%

energy efficiency

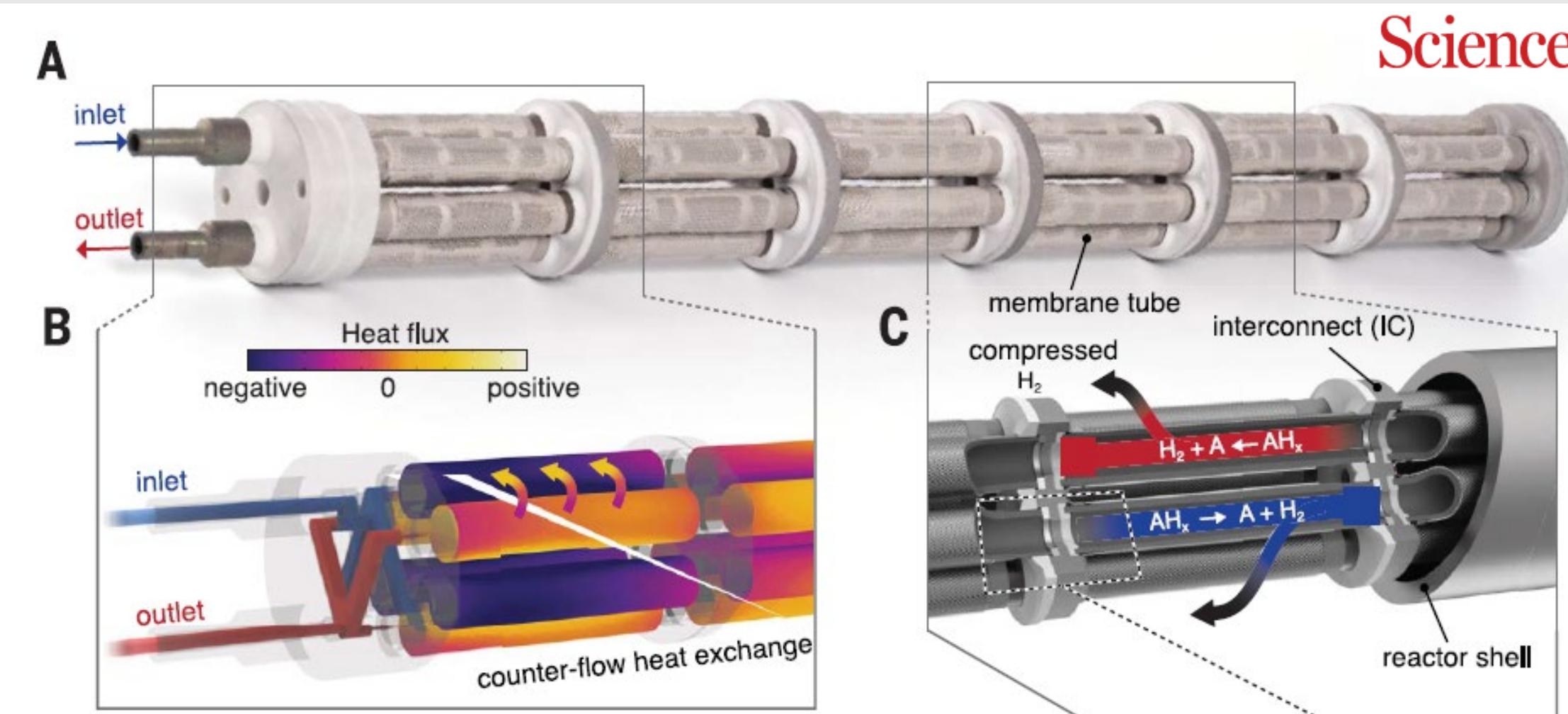
natural gas + electricity → compressed H₂ and CO₂

SOC – Electrified Reforming with Capture

- High current densities ($1\text{-}4 \text{ A/cm}^2$)
- High-pressure operation – 50 bar (150 tested)

D. Clark et. al, Science 376 (2022) p. 390-393

Single-step hydrogen production from NH_3 , CH_4 , and biogas in stacked proton ceramic reactors





Direct electrocatalytic conversion of CO₂ into chemical energy carriers in a co-ionic membrane reactor

AIM: Set-up a technology for direct synthesis of carbon-neutral jet fuels from CO₂ using renewable energy and electrochemical catalytic membrane reactors. Bench-testing targets a 500 W multi-tubular system.

- Single-step electrolysis and one-pot catalytic conversion.
- Operating conditions:
T = 350-450 °C and > 25 bar.



Product:
Jet fuel



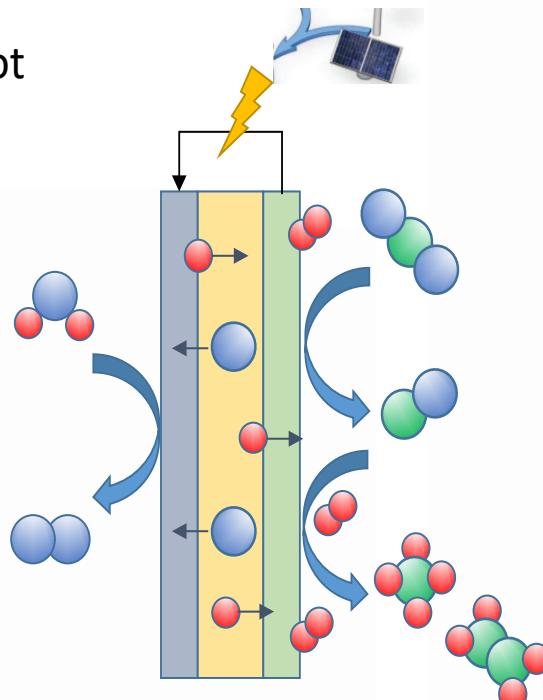
Efficiency:
> 85%



Full integration:
compact sized reactor



Final TRL:
5



PARTNERS



SINTEF



<https://ecocoo.eu>

H2020-LC-SC3-2018-NZE-CC | Duration: May 2019 – May 2023 | EC funding: 3.9 M€

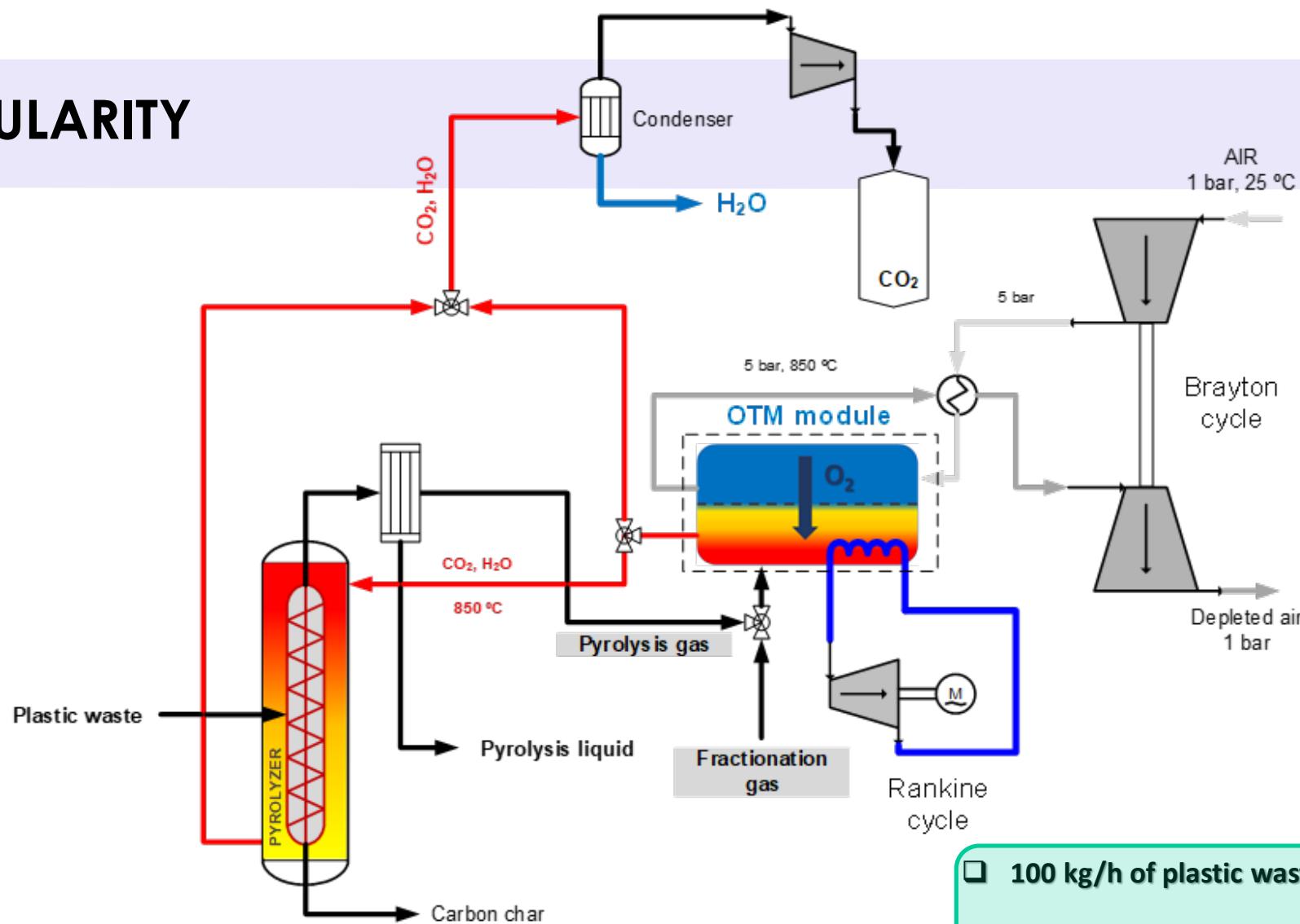
This project has received European Union's Horizon 2020 research and innovation funding under grant agreement № 838077.



Sustainable Process Industry through
Resource and Energy Efficiency



CIRCULARITY



- 100 kg/h of plastic waste is treated.**
- The O₂ demand is around 20.5-26.5 m₃/h.**



FINANCIADO POR LA
UNIÓN EUROPEA
Next Generation EU



GREENRUPTIVE™

MAKING NET ZERO EMISSIONS REAL

PTI+Transener Initiative (Valencia)



PTI+ TRANS-ENER+
Alta Tecnología Clave en la
Transición en el Ciclo Energético

Motivación



Nuestra vocación es ayudar con nuestro know-how
y capacidades experimentales a la industria a
DESARROLLAR SOLUCIONES

- Proyectos piloto de desarrollo e innovación avanzados (TRL-5 a TRL-9)
- Start-ups/spin-off





MISIÓN

- CREAR UN **ECOSISTEMA ABIERTO, INNOVADOR Y COLABORATIVO**
- PROMOVER EL DESARROLLO DE SOLUCIONES Y PRODUCTOS ALINEADOS CON LOS **RETOS INDUSTRIALES** QUE PLANTEA LA ECONOMÍA DE IMPACTO ZERO ENERGÉTICO
- IMPULSAR LA **ESPECIALIZACIÓN** COMO SEÑA DE IDENTIDAD Y PRINCIPAL VENTAJA COMPETITIVA DE UNA INDUSTRIA SOSTENIBLE Y COMPETITIVA
- ATRAER, RETENER Y DESARROLLAR AL MEJOR **TALENTO**

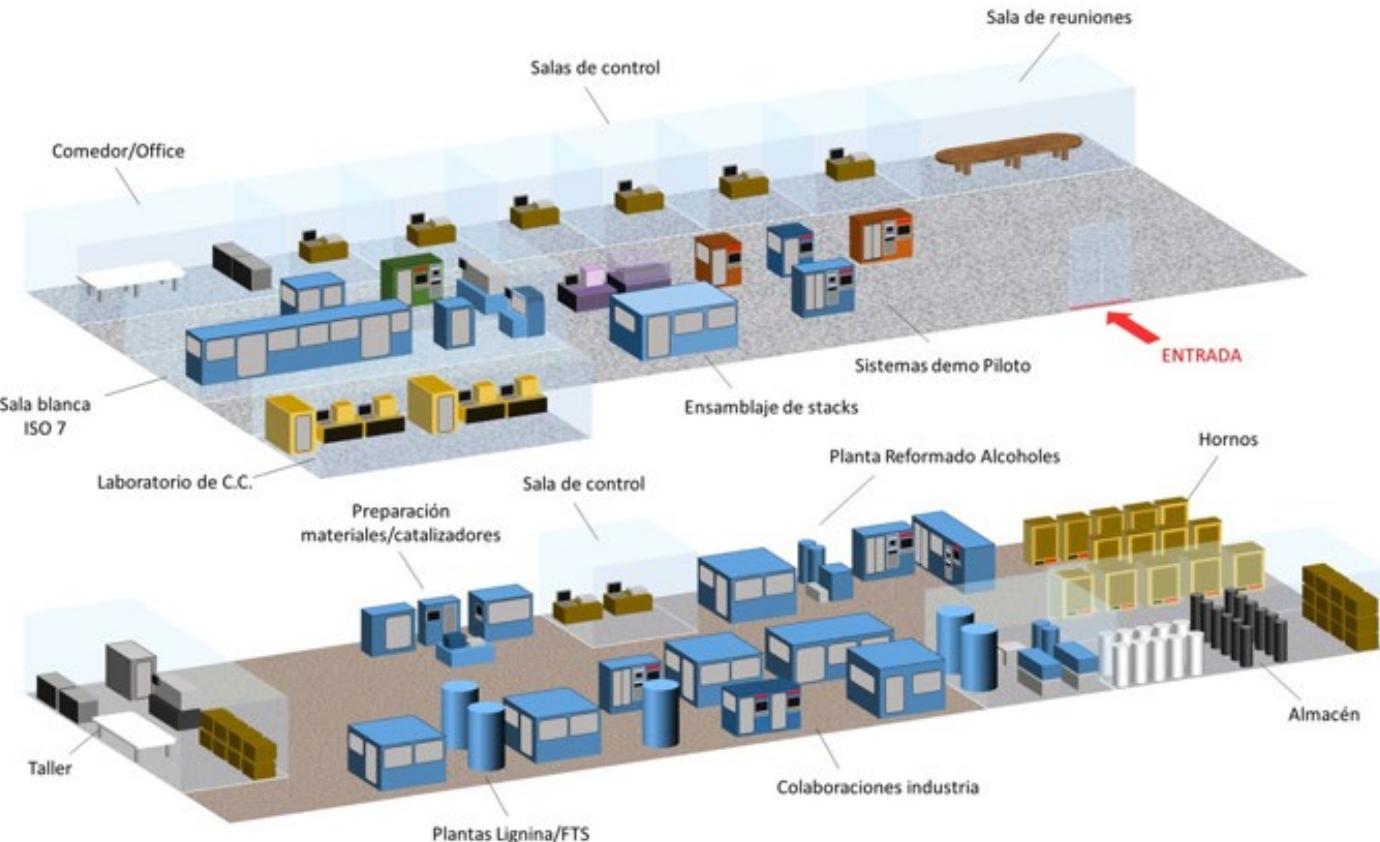
Capacidades clave y Tecnologías

H₂ - e-fuels – Biofuels - Green Steel - Green Ceramics/Cement

Electrificación de procesos - Reactores electroquímicos

Capturas (CCUS, DACC, BECCS) - emisiones negativas

Reciclado de corrientes industriales y residuos bio/agro/RSU/EDARs



Pilotos programados

- Línea piloto de fabricación de celdas SOC/membranas*
- Línea piloto de ensamblado de stacks SOC
- Unidades testeo (BoP) (co-)electrolizadores x 3
- Reformador electrificado NG/biogas/FG con captura x2
- Reformador electrificado etanol con captura x 2
- Unidad testeo Generador de O₂ oxícombustión
- Línea de fabricación de catalizadores
- Piloto generador de H₂ por tecnología microondas
- Reactor de reducción con H₂ en metalurgia
- Reactor de conversión de CO₂ (Fischer-Tropsch)
- Gasificador/pirol. waste/biomas HC o H₂ y captura CO₂
- Unidad de co-electrólisis y síntesis de hidrocarburos

LÍNEAS PILOTO



Capacidades clave: Ejemplos



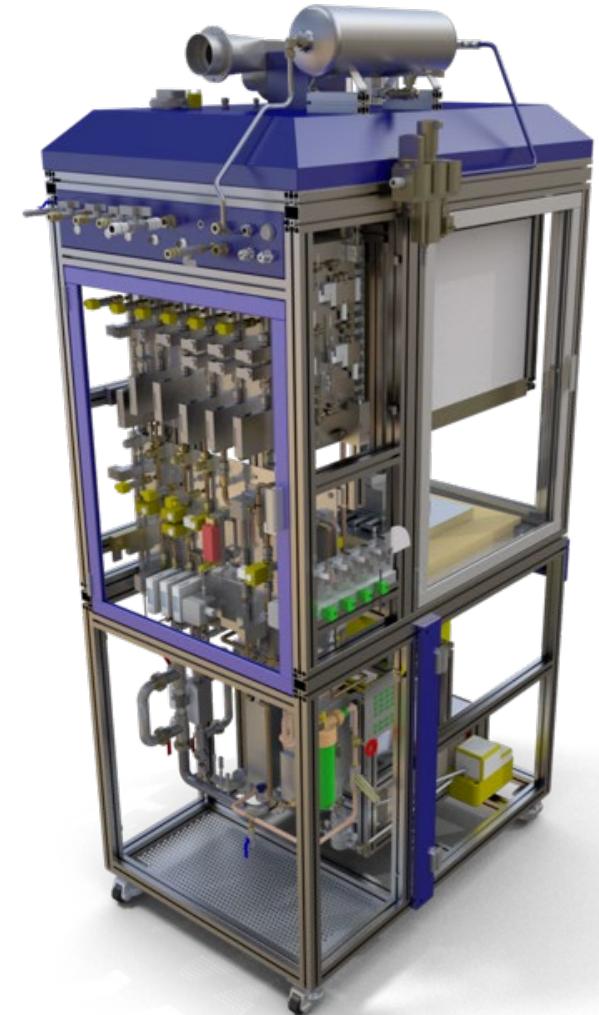
LINEAS DE FABRICACIÓN DE CELDAS
SOC / MEMBRANAS / BATERÍAS



LINEAS DE MONTAJE DE STACK,
MÓDULOS Y REACTORES



BANCOS DE VALIDACIÓN DE
TECNOLOGÍAS CLAVE Y
PILOTOS (TRL5 – TRL9)

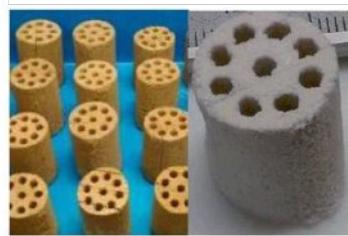


LÍNEAS DE PRODUCCIÓN Y ESCALADO DE
MATERIALES, CATALIZADORES y REACTORES

Selective Laser Sintering



Direct Ink Writing (DIW)



- Proyectos demostrativos (TRL-5 a TRL-9)
- Incubación Start-ups/spin-off
- Formación y especialización
- Campañas de Innovación abierta
- Jornadas específicas
- Value-chain matching (Corporates, EBTs, R&D)
- Consecución de subvenciones específicas
- Inversores

